



# Material Safety Data Sheet

NFPA	HMIS	Personal Protective Equipment
20xy1	Health Hazard 2 Fire Hazard 1 Reactivity 1	
		See Section 15.

Section 1. Chemical Product and Company Identification			Page Number: 1
Common Name/ Trade Name	Ammonium dichromate	Catalog Number(s).	YY022, A1178, A1179
		CAS#	7789-09-5
Manufacturer	SPECTRUM LABORATORY PRODUCTS INC.	RTECS	HX7650000
	14422 S. SAN PEDRO STREET GARDENA, CA 90248	TSCA	TSCA 8(b) inventory: Ammonium dichromate
Commercial Name(s)	Not available.	CI#	Not available.
Synonym	Ammonium Bichromate; Dichromic acid, diammonium salt		ti kan
Chemical Name	Ammonium Dichromate	8	EMERGENCY C (24hr) 800-424-9300
Chemical Family	Not available.	CALL (310) 5	16-8000
Chemical Formula	(NH4)2Cr2O7		
Supplier	SPECTRUM LABORATORY PRODUCTS INC. 14422 S. SAN PEDRO STREET GARDENA, CA 90248		

			Exposure Limits		
Name	CAS#	TWA (mg/m³)	STEL (mg/m³)	CEIL (mg/m³)	% by Weight
1) Ammonium dichromate	7789-09-5	0.05			100

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Potential Acute Health Effects Hazardous in case of skin contact (irritant), of eye contact (irritant), of ingestion, of inhalation. Slightly hazardous in case of

skin contact (permeator). Prolonged exposure may result in skin burns and ulcerations. Over-exposure by inhalation may

cause respiratory irritation. Severe over-exposure can result in death.

Potential Chronic Health

on Ingredients

Hazardous in case of skin contact (permeator). **Effects** 

LD50: Not available. LC50: Not available.

CARCINOGENIC EFFECTS: Classified A1 (Confirmed for human.) by ACGIH, 1 (Proven for human.) by IARC.

MUTAGENIC EFFECTS: Mutagenic for bacteria and/or yeast.

TERATOGENIC EFFECTS: Not available. **DEVELOPMENTAL TOXICITY**: Not available. The substance may be toxic to blood, kidneys, liver.

Repeated or prolonged exposure to the substance can produce target organs damage. Repeated exposure to a highly

toxic material may produce general deterioration of health by an accumulation in one or many human organs.

#### Continued on Next Page

Ammonium dich	romate Page Number: 2
Section 4. First Aid Measures	
Eye Contact	Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Cold water may be used. Get medical attention.
Skin Contact	In case of contact, immediately flush skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Cover the irritated skin with an emollient. Cold water may be used. Wash clothing before reuse. Thoroughly clean shoes before reuse. Get medical attention immediately.
Serious Skin Contact	Wash with a disinfectant soap and cover the contaminated skin with an anti-bacterial cream. Seek immediate medical attention.
Inhalation	If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Get medical attention immediately.
Serious Inhalation	Evacuate the victim to a safe area as soon as possible. Loosen tight clothing such as a collar, tie, belt or waistband. If breathing is difficult, administer oxygen. If the victim is not breathing, perform mouth-to-mouth resuscitation. WARNING: It may be hazardous to the person providing aid to give mouth-to-mouth resuscitation when the inhaled material is toxic, infectious or corrosive. Seek immediate medical attention.
Ingestion	If swallowed, do not induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an unconscious person. Loosen tight clothing such as a collar, tie, belt or waistband. Get medical attention immediately.
Serious Ingestion	Not available.

Section 5. Fire and Explosion Data		
Flammability of the Product	May be combustible at high temperature.	
Auto-Ignition Temperature	190°C (374°F)	
Flash Points	Not available.	
Flammable Limits	Not available.	
Products of Combustion	Some metallic oxides.	
Fire Hazards in Presence of Various Substances	Slightly flammable to flammable in presence of heat. Non-flammable in presence of shocks.	
Explosion Hazards in Presence of Various Substances	Risks of explosion of the product in presence of mechanical impact: Not available. Explosive in presence of open flames and sparks, of heat.	
Fire Fighting Media and Instructions	Oxidizing material. Do not use water jet. Use flooding quantities of water. Avoid contact with organic materials.	
Special Remarks on Fire Hazards	May ignite by friction with carbide. In contact with substances which are readilty oxidized, these can react rapidly enough to cause ignition.	
Special Remarks on Explosion Hazards	Hydrazine is decomposed explosively by chromates. With finely divided oxidizable substances, combustion can be violent. Closed containers readily rupture at decompositon temperature.	

Section 6. Accidental Release Measures		
Small Spill	Use appropriate tools to put the spilled solid in a convenient waste disposal container. If necessary: <b>Neutralize</b> the residue with a dilute solution of sodium carbonate.	
Large Spill	Oxidizing material. Poisonous solid.  Stop leak if without risk. Do not get water inside container. Avoid contact with a combustible material (wood, paper, oil, clothing). Keep substance damp using water spray. Do not touch spilled material. Use water spray to reduce vapors. Prevent entry into sewers, basements or confined areas; dike if needed. Eliminate all ignition sources. Call for assistance on disposal. Neutralize the residue with a dilute solution of sodium carbonate. Be careful that the product is not present at a concentration level above TLV. Check TLV on the MSDS and with local authorities.	

Ammonium dichromate

Ammonium dichromate	Page Number: 3

Section 7. Hai	ndling and Storage
Precautions	Keep locked up Keep away from heat. Keep away from sources of ignition. Keep away from combustible material. Empty containers pose a fire risk, evaporate the residue under a fume hood. Ground all equipment containing material. Do not ingest. Do not breathe dust. Wear suitable protective clothing. In case of insufficient ventilation, wear suitable respiratory equipment. If ingested, seek medical advice immediately and show the container or the label. Avoid contact with skin and eyes. Keep away from incompatibles such as combustible materials, organic materials, acids, alkalis.
Storage	Keep container tightly closed. Keep container in a cool, well-ventilated area. Separate from acids, alkalies, reducing agents and combustibles. See NFPA 43A, Code for the Storage of Liquid and Solid Oxidizers. Do not store above 24°C (75.2°F).

Section 8. Exposure Controls/Personal Protection		
Engineering Controls	Use process enclosures, local exhaust ventilation, or other engineering controls to keep airborne levels below recommended exposure limits. If user operations generate dust, fume or mist, use ventilation to keep exposure to airborne contaminants below the exposure limit.	
Personal Protection	Splash goggles. Lab coat. Dust respirator. Be sure to use an approved/certified respirator or equivalent. Gloves.	
Personal Protection in Case of a Large Spill	Splash goggles. Full suit. Dust respirator. Boots. Gloves. A self contained breathing apparatus should be used to avoid inhalation of the product. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product.	
Exposure Limits	TWA: 0.05 (mg (Cr)/m³) from ACGIH (TLV) [United States]	
- 7	Consult local authorities for acceptable exposure limits.	

Section 9. Physical a	nd Chemical Properties		
Physical state and appearance	Solid. (Crystals solid.)	Odor	Odorless.
Molecular Weight	252.1 g/mole	Taste	Not available.
pH (1% soln/water)	3.95 [Acidic.]	Color	Orange. Red.
Boiling Point	Not available.		
Melting Point	Not available.		
Critical Temperature	Not available.		
Specific Gravity	2.155 (Water = 1)		
Vapor Pressure	Not applicable.		
Vapor Density	8.7(Air = 1)		
Volatility	Not available.		
Odor Threshold	Not available.		
Water/Oil Dist. Coeff.	Not available.		
Ionicity (in Water)	Not available.		
Dispersion Properties	See solubility in water.		
Solubility	Soluble in cold water, hot water. Insoluble in acetone. Soluble in alcohol.		

Ammonium dichro	mate Page Number: 4	
Section 10. Stability and Reactivity Data		
Stability	Unstable.	
Instability Temperature	Not available.	
Conditions of Instability	Excess heat and incompatible materials	
Incompatibility with various substances	Reactive with combustible materials, organic materials, acids, alkalis. Slightly reactive to reactive with moisture.	
Corrosivity	Not available.	
Special Remarks on Reactivity	It is an oxidizing material and combustible solid. Decomposes vigorously with luminescence around 200 C. Decomposes at about 180 C. Decomposition becomes self-sustaining at about 225 C with swelling and evolution of heat and nitrogen.  It is incompatible with combustible, organic, or other readily oxidizable materials: Paper, wood, sulfur, aluminum, plastics, acids, bases, potassium chlorate, sodium nitrite, alcohols, ethylene glycol, water.	
Special Remarks on Corrosivity	Not available.	

Section 11. Toxicological Information					
Routes of Entry	Absorbed through skin. Inhalation. Ingestion.				
Toxicity to Animals	LD50: Not available. LC50: Not available.				
Chronic Effects on Humans	CARCINOGENIC EFFECTS: Classified A1 (Confirmed for human.) by ACGIH, 1 (Proven for human.) by IARC.  MUTAGENIC EFFECTS: Mutagenic for bacteria and/or yeast.  May cause damage to the following organs: blood, kidneys, liver.				
Other Toxic Effects on Humans	Hazardous in case of skin contact (irritant), of ingestion, of inhalation.  Slightly hazardous in case of skin contact (permeator).				
Special Remarks on Toxicity to Animals	Lowest Published Lethal Dose LDL [Child] - Route: Oral; Dose: 99 mg/kg				
Special Remarks on Chronic Effects on Humans	Not available.				
Special Remarks on other Toxic Effects on Humans	Acute Potential Health Effects: Skin: Causes skin irritation. May be absorbed through skin Eyes: Causes eye irritation. Inhalation: Causes respiratory tract and mucous membrane irritation. It may destructive to the tissues of the mucous membranes Ingestion: May cause intense gastrointestinal tract irritation. May affect blood, Kidneys (urinary system), and liver. Symptoms of acute poisoning may include ulceration and corrosion, epigastric pain, nausea, vomiting, diarrhea, vertigo, fever, muscle cramps, hemorrhagic diathesis, toxic nephritis, renal failure, intravascular hemolysis, circulatory collapse, liver damage, peripheral vascular collapse, acute multisystem shock and coma, and even death depending on the dose. Chronic Potential Health Effects: Chronic poisoning usually results from inhalation or skin contact. May affect the blood, kidneys and liver. Signs and symptoms may include lacrimation, dermatitis, penetrating ulcers, perforation of nasal septum, pulmonary edema, congestion, chronic rhinitis, polyps of the upper respiratory tract, inflammation of lung, emphysema, tracheitis, bronchitis, pharyngitis, adhesions of the diaphragm, inflammation of larynx, conjunctivitis, loss of appetite, nausea, vomiting, inflammation of liver or even acute hepatitis with jaundice, respiratory irritations, leukocytosis, leukopenia, monocytosis, and eosinophilia.				

Polymerization

Will not occur.

Ammonium dichromate		Page Number: 5	
Section 12. Ecological Information			
Ecotoxicity	Not available.		
BOD5 and COD	Not available.		
Products of Biodegradation	of Biodegradation Possibly hazardous short/long term degradation products are to be expected.		
Toxicity of the Products of Biodegradation			

# Section 13. Disposal Considerations

Waste Disposal Waste must be disposed of in accordance with federal, state and local environmental

control regulations.

Not available.

## Section 14. Transport Information

**DOT Classification** CLASS 5.1: Oxidizing material.

Identification : Ammonium dichromate UNNA: 1439 PG: II

Not available.

Special Provisions for

Special Remarks on the

Products of Biodegradation

Transport

DOT (Pictograms)



#### Section 15. Other Regulatory Information and Pictograms

Federal and State Regulations California prop. 65: This product contains the following ingredients for which the State of California has found to cause cancer, birth defects or other reproductive harm, which would require a warning under the statute:

Ammonium dichromate

California prop. 65: This product contains the following ingredients for which the State of California has found to

cause cancer which would require a warning under the statute: Ammonium dichromate

Connecticut hazardous material survey.: Ammonium dichromate

Illinois chemical safety act: Ammonium dichromate

New York release reporting list: Ammonium dichromate

Rhode Island RTK hazardous substances: Ammonium dichromate

Pennsylvania RTK: Ammonium dichromate Massachusetts RTK: Ammonium dichromate

Massachusetts spill list: Ammonium dichromate

New Jersey: Ammonium dichromate

New Jersey spill list: Ammonium dichromate

Louisiana spill reporting: Ammonium dichromate

California Director's List of Hazardous Substances: Ammonium dichromate

TSCA 8(b) inventory: Ammonium dichromate

CERCLA: Hazardous substances.: Ammonium dichromate: 10 lbs. (4.536 kg)

California Proposition 65 Warnings California prop. 65: This product contains the following ingredients for which the State of California has found

to cause cancer which would require a warning under the statute: Ammonium dichromate

Other Regulations OSHA: Hazardous by definition of Hazard Communication Standard (29 CFR 1910.1200).

EINECS: This product is on the European Inventory of Existing Commercial Chemical Substances.

Inventory lists for China, Japan, Korea, and Philippines.

#### Continued on Next Page

## Ammonium dichromate Page Number: 6 Other Classifications WHMIS (Canada) CLASS C: Oxidizing material. CLASS D-1A: Material causing immediate and serious toxic effects (VERY TOXIC). CLASS D-2A: Material causing other toxic effects (VERY TOXIC). DSCL (EEC) R8- Contact with combustible material S2- Keep out of the reach of children. may cause fire. S46- If swallowed, seek medical advice R21- Harmful in contact with skin. immediately and show this container or label. R25- Toxic if swallowed. S53- Avoid exposure - obtain special R26- Very toxic by inhalation. instructions before use. R37/38- Irritating to respiratory system R41- Risk of serious damage to eyes. R43- May cause sensitization by skin contact. R45- May cause cancer. R46- May cause heritable genetic damage. Health Hazard HMIS (U.S.A.) 2 **National Fire Protection** Flammability Association (U.S.A.) Fire Hazard Health Reactivity Reactivity 1 Personal Protection Specific hazard $\mathbf{E}$ WHMIS (Canada) (Pictograms) DSCL (Europe) (Pictograms) TDG (Canada) (Pictograms) ADR (Europe) (Pictograms) Protective Equipment Gloves. Lab coat. Dust respirator. Be sure to use an approved/certified respirator or equivalent. Wear appropriate respirator when ventilation is inadequate. Splash goggles.

	Ammonium dichromate	Page Number: 7
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Section 16. Other Information		
MSDS Code	A5050	
References	Not available.	
Other Special Considerations	Not available.	
Validated by Sonia Owen on 8/11/2006.		Verified by Sonia Owen. Printed 9/8/2006.
CALL (210) 51( 9000		

CALL (310) 516-8000

#### Notice to Reader

All chemicals may pose unknown hazards and should be used with caution. This Material Safety Data Sheet (MSDS) applies only to the material as packaged. If this product is combined with other materials, deteriorates, or becomes contaminated, it may pose hazards not mentioned in this MSDS. It shall be the user's responsibility to develop proper methods of handling and personal protection based on the actual conditions of use. While this MSDS is based on technical data judged to be reliable, Spectrum Quality Products, Inc. assumes no responsibility for the completeness or accuracy of the information contained herein.